

Available online at www.sciencedirect.com

Tetrahedron

Tetrahedron 63 (2007) 2170–2181

Synthesis of biantennary LacNAc-linked O-glycan (core 4) and glycopeptide thioester by benzyl protection strategy: rapid zinc reduction of GlcNTCA to GlcNAc by microwave irradiation

Akiharu Ueki, Yuko Nakahara, Hironobu Hojo* and Yoshiaki Nakahara*

Department of Applied Biochemistry, Institute of Glycotechnology, Tokai University, 1117 Kitakaname, Hiratsuka-shi, Kanagawa 259-1292, Japan

> Received 1 December 2006; revised 22 December 2006; accepted 25 December 2006 Available online 22 January 2007

Abstract—A synthetic method for the core 4 O-glycan-linked Ser and Thr was developed. Highly stereoselective 3-O- and 6-O-glycosylation was achieved by using two distinctively protected N-trichloroacetyllactosaminyl fluorides (3 and 12). Microwave-assisted Zn reduction rapidly and efficiently converted N-trichloroacetylglucosamine (GlcNTCA) to N-acetylglucosamine (GlcNAc). In order to demonstrate the usefulness of the protected core $4 O$ -glycan a segment $(Gly^{34}-Gly^{58})$ of emmprin (extracellular matrix metalloproteinase inducer), a cancer metastasis-related glycoprotein, was synthesized by the solid-phase method, utilizing the pentasaccharyl Thr (2) to introduce an O-glycan in place of the native N -glycan at Asn⁴⁴.

© 2007 Elsevier Ltd. All rights reserved.

1. Introduction

Mucins and their O-glycans are of great importance and interest in a number of biological processes. Aberrant features of neoplastic mucins, such as overexpression and altered glycosylation, have attracted particular attention in connec-tion with metastasis.^{[1](#page-11-0)} However, only limited knowledge of the biological roles of the alteration in mucins has been obtained so far. By considering the inaccessibility of a homogeneous mucin sample from natural sources, we have studied a synthetic approach to the glycoproteins with O-glycan, and recently established an original protocol using the benzylprotected glycoamino acid building blocks in solid-phase glycopeptide synthesis.^{[2](#page-11-0)} In a previous study, we have synthesized the core 3 and core 6 oligosaccharides by glycosylating either the 3- or 6-hydroxyl group of the core N-acetylgalactosamine precursor with an N-trichloroacetyllactosaminyl glycosyl donor of high reactivity and β -selectivity. Usefulness of the synthetic O-glycan building blocks was demonstrated by the synthesis of MUC2 and MUC6 related glycopeptides.^{[3](#page-11-0)}

The N-acetylglucosaminyl substitution at both 3- and 6 position gives another core class O-glycan, known as core

4, which has been identified in the oligosaccharides from human bronchial mucins of cystic fibrosis patients, 4 secreted mucins of a human colonic cancer cell line,^{[5](#page-11-0)} human meconium mucins, 6 and sheep gastric mucins.^{[7](#page-11-0)} The core 4 oligosaccharides bearing the N-acetyllactosamine branches are of particular interest regarding an unanswered question, whether their physical, structural, and biological properties are different from those of the complex-type N-glycan as well as those of the core 2 O-glycan having an extension of N-acetyllactosamine to the core galactose residue.^{[6,8](#page-11-0)} To this end our investigations were directed to the synthesis of a glycopeptide with core 4 O-glycan. In this paper, we describe preparation of the core 4 glycoserine and glycothreonine building blocks, 1 and 2, and performance of the solid-phase glycopeptide synthesis with 2 according to the established protocol.^{[9](#page-11-0)}

2. Synthesis of the building blocks 1 and 2

We first attempted selective di-O-glycosylation of 3,4,6- O-unmasked GalN₃–Thr derivative 4^{10} 4^{10} 4^{10} with known N-trichloroacetyllactosaminyl fluoride 3^{11} 3^{11} 3^{11} (2.2 equiv) by using $\text{Cp}_2\text{ZrCl}_2/\text{AgClO}_4$ as the promoter^{[12](#page-11-0)} in CH_2Cl_2 at -15 °C, since the 4-hydroxyl group of the $GalN₃$ residue was hardly glycosylated in many cases (Fig. 1).^{[3,10,11](#page-11-0)} This simple strategy, however, was unsuccessful and gave a complex mixture of a heptasaccharide and three pentasaccharides each in 5–14% yield after consuming the glycosyl fluoride for 3 h. As the second attempt, we reacted 3 and 6-O-silylated

Keywords: Glycopeptide; Core 4 O-glycan; Solid-phase synthesis; Microwave reaction.

Corresponding authors. Tel./fax: +81 463 50 2075; e-mail addresses: hojo@keyaki.cc.u-tokai.ac.jp; yonak@keyaki.cc.u-tokai.ac.jp

^{0040–4020/\$ -} see front matter © 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2006.12.088

Figure 1. Structures of the protected core 4 pentasaccharyl Ser/Thr (1 and 2) and the known intermediates (3–7).

acceptor 5[10](#page-11-0) in expectation of attaining selective 3-O-glycosylation. But a pentasaccharide (12%) derived by 3,4-di-O-glycosylation was produced along with the desired trisaccharide (13%). In this reaction, an additional complication arose from the departure of the acid-labile silyl group of 5 under the reaction conditions. Thus, we were convinced that the side reaction on the 4-hydroxyl group was unavoidable, when this reactive glycosyl donor was used with the 3,4-unprotected glycosyl acceptors. In order to secure mono 3-O-glycosylation, we decided to use 4,6-O-benzylidene GalN₃–Ser/Thr derivatives, 6^{13} 6^{13} 6^{13} and 7^{10} 7^{10} 7^{10} as the glycosyl acceptors, and instead needed a benzylidene group-free glycosyl donor that allowed selective deprotection of the 6-O

position of the $GalN₃$ residue at a later stage. Thus, a perbenzylated N-trichloroacetyllactosaminyl fluoride was synthesized as an alternative to glycosyl donor 3. Known lactosamine derivative 8^{14} 8^{14} 8^{14} was heated with ethylenediamine in n -BuOH to remove the N -phthaloyl group (Scheme 1). The resulting amine 9 was reacted with trichloroacetyl chloride in pyridine to give 10 (81% in two steps). Desilylation of 10 with $n-Bu₄NF$ in THF in the presence of excess AcOH afforded hemiacetal 11, which upon treatment with $Et₂NSF₃$ gave fluoride 12 (82% in two steps) as a mixture of anomers $(\alpha/\beta=19/1)$. Fluoride 12 seemed more reactive than 3, and reacted with glycosyl serine 6 within 0.5 h by activation with $\text{Cp}_2\text{Zr}(\text{ClO}_4)_2$ at -15 °C to afford trisaccharide 13 as

: $R^1 = H$, R^2 , $R^3 = -CH(Ph)$ -: R^1 = CH₃, R^2 , R^3 = -CH(Ph)-: $R^1 = H$, R^2 , $R^3 = H$: R^1 = CH₃, R^2 , R^3 = H

Scheme 1. Synthesis of hexabenzylated glycosyl fluoride 12 and trisaccharyl serine/threonine, 15 and 16. Reaction conditions: (a) 1,2-diaminoethane, n-BuOH, 90 °C, 2 days, 96%; (b) trichloroacetyl chloride, pyridine, 0 °C, 1.5 h, 84%; (c) n-Bu₄NF, AcOH, THF, room temperature, overnight, 89%; (d) diethylaminosulfur trifluoride, THF, 0 °C, 1 h, 92%; (e) 6 or 7, Cp₂ZrCl₂, AgClO₄, CH₂Cl₂, -15 °C, 1 h, **13** (75%), **14** (80%); (f) 80% aq TFA, CH₂Cl₂, 94% (**15**), 83% (**16**).

the sole coupling product in 75% yield. Similarly, trisaccharyl threonine 14 was obtained from 7 and 12 in 80% yield. Stereochemistry of the newly generated linkage was evidenced by the $^{1}J_{\text{CH}}$ value (168.4 Hz for 13 and 165.9 Hz for 14). Removal of the benzylidene group of 13 and 14 with 80% TFA led to 4,6-dihydroxy compounds, 15 and 16.

Selective 6-O-glycosylation was first tested with glycosyl donor 12 and diol 16. The reaction, however, again gave an unsatisfactory result. Two pentasaccharides were produced, when donor 12 and excess acceptor 16 (12/ **16**=0.83) were reacted in CH₂Cl₂ at -15 °C for 1 h. 4-O-Glycosylated product 18 was obtained in a considerable amount (30%) along with the desired 17 (44%). The structures of 17 and 18 were determined after acetylation of the product samples (Fig. 2). The characteristic lower field shift of the H-4 signal (GalN₃) was observed at 5.36 ppm for the acetate derivative of 17. The altered condition with the

BnO₂OBn

O_{Bn}

lowered temperature or with other solvents was ineffective for suppressing the 4-O-glycosylation. On the other hand, we had achieved 6-O-selective glycosylation to the desired extent by using glycosyl donor 3 in the synthesis of core 2 and core 6. Thus, condensation of 3 and 16 was next investigated. When 3 was reacted with excess $16 \left(\frac{3}{16} = 0.83 \right)$ at -20 °C for 2.5 h, the desired pentasaccharide 20 was produced in a moderate yield (40%) and acceptor 16 (54%) was recovered. In contrast, the use of excess donor $(3/$ $16=1.2$) promoted better conversion of 16, but resulted in formation of the byproducts, 22 and 24. Through several attempts to reduce the side reactions, an improved yield (71%) of 20 was attained under the reaction of 3 and 16 $(3/16=1.1)$ at -40 to -15 °C for 2 h. However, the formation of 22 (9%) and 24 (8%) could not be suppressed. Structures of the products were assigned by ${}^{1}H$ and ${}^{13}C$ NMR. The same reaction conditions were applied to the condensation of 3 and serine derivative 15 to produce 19 (71%) , 21 (6%) , and 23 (4%) . With no further optimization, synthesis was forwarded using

O

OH O

 N_3 ¹

 Fm ocNH \sim CO₂All

 H_3C

O

21: R = H **22**: $R = CH_3$

the obtained pentasaccharides 19 and 20. In the previous studies, we have accomplished dechlorination of the trichloroacetamido group and simultaneous reduction of the azido group with large excess of Zn and AcOH. Despite giving a fairly good yield, the tedious procedure of the reaction was a drawback. Intermittent additions of Zn powder and AcOH to the reactant in CH_2Cl_2 were necessary at 3–5 h intervals during 1–2 days until completion of the dechlorination. In addition, $Zn(OAc)$ ₂ resulted in a large quantity, which gave some difficulty in the following acetylation of the crude reduction product. In recent years, acceleration of reactions by microwave irradiation has received considerable attention.[15](#page-11-0) The apparent effectiveness reported for the reactions in a heterogeneous system prompted us to investigate the Zn reduction under microwave. We examined several conditions by varying the amount of the reductant, microwave power, reaction time, and solvent with a model monosaccharide as well as 19. After optimization, the best result was obtained with a reduced amount of Zn by microwave heating gently at refluxing temperature in EtOAc for 30 min. The reduction products derived from 19 and 20 were then acetylated with Ac_2O in $CH_2Cl_2/MeOH$ to give 25 (81%) and 26 (80%), respectively. To fulfill the synthesis of the building blocks suitable for the solid-phase synthesis of glycopeptide, 25 and 26 were converted to carboxylic acids 1 (93%) and 2 (92%) by deallylation with $Pd(PPh_3)_4$ and 5,5-dimethyl-1,3-cyclohexanedione in THF.

3. Solid-phase synthesis of glycopeptide

As a suitable platform that allows us to test the effectiveness of the benzyl and benzylidene-protected core 4 building blocks in the solid-phase synthesis, the chosen model glycopeptide was a glycosylated pentacosapeptide thioester that mimics a segment $\left(\frac{Gly^{34}-Gly^{58}}{\sqrt{Gly^{58}}}\right)$ of the extracellular matrix metalloproteinase inducer (emmprin). Emmprin is produced by tumor cells and is thought to function in the early stage of metastasis. The native emmprin carries an N-linked oligosaccharide at the $Asn⁴⁴$ site, and the first Ig domain $(Gly³⁴-Met⁹⁴)$ (Fig. 3) including the N-glycan is responsible for the inducer activity. Recently, we have established an efficient synthetic route to the first Ig domain and its extended peptide bearing chitobiosyl or core pentasaccharyl motif of the N-glycan by taking advantage of the thioester method of segment condensation.[16](#page-11-0) In the present study, building block 2 was introduced into the significant segment in place of the N -glycan at the Asn⁴⁴ position. The synthetic imitation would be helpful to explore the structural requirement of the oligosaccharide attached to the bioactive Ig domain, since the biantennary LacNAc in the core 4 O-glycan pentasaccharide resembles the peripheral structure of the complex-type N-glycan.

Glycopentacosapeptide 27, which carries a thioester functionality necessary for the segment condensation in the Ig domain synthesis, was synthesized from commercial Fmoc-CLEAR-amide resin [\(Scheme 2\)](#page-4-0). Attachment of Fmoc– $Gly-SCH₂CH₂CO₂H$ to the *N*-deprotected resin was followed by removal of the Fmoc group using a cocktail of 1-methylpyrrolidine (25%), hexamethyleneimine (2%), and 1-hydroxybenzotriazole (2%) in NMP/DMSO $(1:1).$ ^{[17](#page-11-0)} According to the reported procedure, $16b$ the tripeptide-resin was prepared by coupling first with N^6 -benzyloxycarbonyl- N^2 -triisopropylsilyloxycarbonyl-lysine pentafluorophenyl ester [(Tsoc)–Lys(Z)–OPfp] and then with Fmoc–Leu–F in the presence of tetrabutylammonium fluoride. This twostep protocol is indispensable to get rid of the side reaction caused by a diketopiperazine formation in the early stage of thioester synthesis. Further elongation with 11 amino acids was performed with an automated peptide synthesizer using Fmoc amino acids activated with HBTU, HOBt, and DIEA. All the N-deprotection steps were carried out with the above cocktail under the automated program. Benzyloxycarbonyl group (Z), tert-butoxycarbonyl group (Boc), 2,2,4,6,7-pentamethyldihydrobenzofuran-5-sulfonyl group (Pbf), and triphenylmethyl group (Trt) were employed for protection of the functional side chain amino groups of

Figure 3. Structures of the emmprin Ig domain (Gly34-Met94) and synthetic glycopeptide 27.

Scheme 2. Solid-phase synthesis of glycopeptide 27. Reaction conditions: (a) 20% piperidine/NMP; (b) Fmoc–Gly–SCH₂CH₂CO₂H, DCC/HOBt/NMP; (c) 1-methylpyrrolidine/hexamethyleneimine/1-hydroxybenzotriazole/NMP/DMSO; (d) Tsoc–Lys(Z)–OPfp; (e) Fmoc–Leu–F, TBAF, CH2Cl2; (f) Fmocamino acid, HBTU, HOBt, DIEA, ABI 433A peptide synthesizer, FastMoc program; (g) 2, DCC/HOBt/NMP, 50 °C, 3 h; (h) Fmoc amino acid, DCC/ HOBt/NMP, 50 °C; (i) TFA/phenol/thioanisole/EDT/H₂O, room temperature, 2 h; (j) DMS/TFA/m-cresol, -15 °C, then TfOH, 2 h.

Lys, Trp, Arg, and His, respectively, whereas the tert-butyl group ('Bu) was used for masking the hydroxyl groups of Ser and Thr, and the carboxyl groups of Glu and Asp residues. These protecting groups are readily removed under acidic conditions. The N-deprotected tetradecapeptide-resin (25 μ mol) was coupled with 2 (2 equiv) by manual operation using DCC and HOBt at 50° C for 3 h in a polypropylene vessel with stirring by a vortex mixer. Among the tested conditions including other condensing agents, the combination of DCC and HOBt gave the most acceptable coupling efficiency. The coupling was monitored by HPLC and MS analyses of the glycopeptide detached from the resin sample. The glycopeptide chain was further elongated manually to complete the pentacosapeptide synthesis. For persistent protection of the thiol group of Cys, an acetamidomethyl group (Acm) was employed. According to our standard protocol for isolation of synthetic glycopeptide, a series of deprotection procedures were conducted. The resin was treated with reagent K (aq CF_3CO_2H , thioanisole, 1,2-ethanedithiol, and phenol)[18](#page-11-0) for 2 h to split the glycopeptide from the resin. The crude product was thoroughly debenzylated under the 'low-acidity TfOH' conditions^{[19](#page-11-0)} at -15 °C for 2 h and then purified by HPLC. Figure 4 shows that the desired product 27 was obtained as the major product (peak 1). The structure of the synthesized glycopeptide thioester was evidenced by MALDI TOF MS $[m/z]$ 3764.95 (M+H⁺)]. A byproduct eluted in the less mobile fraction (peak 2) was assigned as a glycopeptide lacking a LacNAc moiety [m/z 3399.62 (M+H⁺)]. Although the resulting glycan structure of the latter glycopeptide has not been elucidated, it is probable that an acid-labile 6-O-glycosidic linkage might have been cleaved as observed in the synthetic glycopeptide carrying core 6 O-glycans.^{[3](#page-11-0)} The overall yield of purified 27 was deduced to be 2.2% from the data of amino acid analysis of the acid-hydrolyzed sample, and the value is comparable to that obtained previously for the synthetic N-glycan-linked glycopeptide.^{[16b](#page-11-0)} The LacNAc-missing glycopeptide was also isolated in 1.3% overall yield. The isolated 27 (total

Figure 4. HPLC profiles of synthetic glycopeptide 27 (a) and isolated 27 (b). Conditions: column, Mightysil RP-18, 4.6×150 mm (5 µm); eluant A, distilled water containing 0.1% TFA; eluant B, acetonitrile containing 0.1% TFA; flow rate, 1 ml/min.

 0.6μ mol, [Fig. 4](#page-4-0)) is a useful intermediate to synthesize an O-glycan analog of the emmprin first Ig domain, through N-tert-butoxycarbonylation, condensation with the known segment $\left(\text{Gly}^{59}\text{-Met}^{54}\right)$, and disulfide formation.

In summary, the biantennary LacNAc-linked O-glycosyl Ser and Thr were synthesized both in a benzyl and benzylideneprotected form by using two distinctively protected LacNTCA glycosyl donors. The dechlorination of the trichloroacetamido group by Zn/AcOH reduction was greatly accelerated by employing microwave irradiation. By the solid-phase synthetic procedure suited for the labile Fmoc peptide thioester, the glycothreonine derivative was successfully introduced into a glycopeptide, which mimics the glycosylated segment in the metastasis-related emmprin first Ig domain. The benzyl-protecting groups of the glycan moiety were readily removed under the low-acidity TfOH conditions to complete the desired glycopeptide thioester. Synthesis toward the 61 amino acid structure for the Ig domain and bioassay with the glycan-mutated sample will be reported in due course.

4. Experimental

4.1. General

Optical rotation values were determined with a Jasco DIP-370 polarimeter at 20 ± 2 °C for solutions in CHCl₃, unless noted otherwise. Column chromatography was performed on silica gel PSQ 100B (Fuji Silysia). TLC and HPTLC were performed on silica gel 60 F_{254} (E. Merck). ¹H and ¹³C NMR spectra were recorded with a Jeol AL400 spectrometer $[$ ¹H (400 MHz) and ¹³C (100 MHz)]. Chemical shifts are expressed in parts per million downfield from the signal for internal Me₄Si for solutions in CDCl₃. For assignment of the signals of pentasaccharides, the reducing terminal residue GalNAc is described as a. GlcNAc and Gal residues in the 3-O-substituent of GalNAc are described as b and c, respectively, while GlcNAc and Gal residues in the 6(or 4)-O-substituent of GalNAc are shown as d and e. For heptasaccharides 23 and 24, GlcNAc and Gal residues in the 6-O-substituent of GalNAc are described as f and g, respectively. MALDI TOF mass spectra were obtained with a PerSeptive Voyager-DE PRO spectrometer (2,5-dihydroxybenzoic acid was used as a matrix). Microwave irradiation was carried out with a CEM Discover microwave reactor. Automated solid-phase peptide synthesis was performed with Applied Biosystems Model 433A peptide synthesizer. Manual solid-phase reactions were undertaken in capped polypropylene test tubes equipped with a filter and threeway stopcock by stirring with an EYELA CM-1000 vortex mixer. HPLC was performed with Mightysil RP-18 $(4.6\times150$ mm for analysis and 10×250 mm for preparation, Kanto Chemical Co.). Amino acids were analyzed by a Hitachi L-8500 amino acid analyzer. Fmoc-CLEAR-amide resin was purchased from Peptide International Inc. Due to the thermolabile azide group, N analysis values of compounds 13, 14, and 15 showed some inconsistency. The yield of glycopeptide was determined by amino acid analysis after a measured volume from the whole sample solution was hydrolyzed in a sealed tube with 20% HCl and 0.5% phenol at 150° C for 2 h.

4.1.1. tert-Butyldiphenylsilyl 2,3,4,6-tetra-O-benzyl-b p -galactopyranosyl- $(1\rightarrow 4)$ -2-amino-3,6-di-O-benzyl-2deoxy- β -D-glucopyranoside 9. A mixture of 8 (703 mg, 0.56 mmol) and 1,2-diaminoethane (0.75 ml, 11.2 mmol) in *n*-BuOH (25 ml) was heated at 90 °C with stirring for 2 days under Ar, before being concentrated in vacuo. The residue was extracted with CHCl₃, successively washed with water and brine, dried over $Na₂SO₄$, and concentrated in vacuo. The crude product was chromatographed on silica gel with CHCl₃ to give 9 (603 mg, 96%). $[\alpha]_D - 5.3$ (*c* 1.0). R_f 0.43 (1:1 hexane/EtOAc). ¹H NMR: δ 7.71–7.67, 7.49– 7.10 (m, 40H, Ar), 5.16 (d, 1H, J=10.5 Hz, -CH₂Ph), 4.97 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.76–4.69 (m, 4H, 4 \times $-CH_2Ph$, 4.58–4.47 (m, 2H, 2× $-CH_2Ph$), 4.45 (d, 1H, $J=7.8$ Hz, H-1b), 4.39 (d, 1H, $J=12.0$ Hz, $-CH₂Ph$), 4.37 (d, 1H, $J=7.1$ Hz, H-1a), 4.36 (d, 1H, $J=12.2$ Hz, $-CH_2Ph$, 4.26 (d, 1H, J=11.7 Hz, $-CH_2Ph$), 4.20 (d, 1H, $J=12.2$ Hz, $-CH_2Ph$, 4.01 (t, 1H, $J=9.3$ Hz, H-4a), 3.91 (d, 1H, $J=2.4$ Hz, H-4b), 3.72 (t, 1H, $J=8.1$ Hz, H-2b), 3.67 (dd, 1H, $J=2.9$, 11.2 Hz, H-6a), 3.53 (m, 1H, H-6a), 3.43–3.37 (m, 3H, H-3b, H-5b, and H-6b), 3.26–3.21 (m, 2H, H-3a and H-6b), 2.91–2.98 (m, 2H, H-2a and H-5a), 1.10 (s, 9H, 'Bu). ¹³C NMR: δ 102.4 (C-1b), 98.8 (C-1a). MALDI TOF MS calcd for $C_{70}H_{77}NO_{10}Si$ m/z: 1119.53. Found: 1142.75 (M+Na⁺). Anal. Calcd for $C_{70}H_{77}NO_{10}Si$: C, 75.04; H, 6.93; N, 1.25. Found: C, 75.01; H, 6.83; N, 1.14.

4.1.2. tert-Butyldiphenylsilyl 2,3,4,6-tetra-O-benzyl-b p -galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy- 2 -trichloroacetamido- β -D-glucopyranoside 10. To an ice-cooled solution of 9 (3.43 g, 3.06 mmol) in pyridine (40 ml) was added CCl₃COCl (513 μ l, 4.06 mmol). The mixture was stirred for 1.5 h at 0° C, and then concentrated in vacuo. The residue was extracted with CHCl₃, washed successively with water and brine, dried over $Na₂SO₄$, and concentrated invacuo. Chromatography of the crude on silica gel with hexane/EtOAc (4:1) afforded 10 (3.23 g, 84%). $[\alpha]_D$ -3.3 (c 1). R_f 0.66 (1:1 hexane/EtOAc). ¹H NMR: δ 7.69– 7.62, 7.39–7.10 (m, 40H, Ar), 6.88 (d, 1H, $J=7.8$ Hz, $-NH$), 4.95 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$), 4.94 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.89 (d, 1H, $J=7.1$ Hz, H-1a), 4.76– 4.66 (m, 4H, $4 \times -CH_2Ph$), 4.54 (d, 1H, J=10.5 Hz, $-CH_2Ph$, 4.52 (d, 1H, J=11.5 Hz, $-CH_2Ph$), 4.44 (d, 1H, $J=7.6$ Hz, H-1b), 4.37 (d, 1H, $J=12.2$ Hz, $-CH₂Ph$), 4.33 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$), 4.23 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$, 4.22 (d, 1H, J=12.0 Hz, $-CH_2Ph$), 4.04 (t, 1H, $J=8.8$ Hz, H-4a), 3.89 (d, 1H, $J=2.4$ Hz, H-4b), 3.83 (t, 1H, J=7.8 Hz, H-3a), 3.77 (dd, 1H, J=7.8, 9.8 Hz, H-2b), 3.71 (dd, 1H, $J=7.8$, 9.8 Hz, H-2b), 3.66 (dd, 1H, $J=3.4$, 11.2 Hz, H-6a), 3.41 (dd, 1H, $J=2.9$, 9.8 Hz, H-3b), 3.40– 3.37 (m, 1H, H-5b), 3.33–3.29 (m, 2H, H-6a and H-6b), 3.07–3.05 (m, 1H, H-5a), 1.05 (s, 9H, ^t Bu). 13C NMR: δ 161.2 (Cl₃CCONH), 102.7 (C-1b), 94.8 (C-1a), 92.5 $(-CCl₃)$. MALDI TOF MS calcd for $C₇₂H₇₆Cl₃NO₁₁Si m/z$: 1263.43. Found: 1286.26 (M+Na⁺). Anal. Calcd for $C_{72}H_{76}Cl_3NO_{11}Si$: C, 68.32; H, 6.05; N, 1.11. Found: C, 68.51; H, 5.84; N, 1.01.

4.1.3. 2,3,4,6-Tetra-*O*-benzyl-β-D-galactopyranosyl- $(1 \rightarrow$ 4)-3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-a-Dglucopyranose 11. To an ice-cooled mixture of 10 (2.03 g, 1.60 mmol) and AcOH (912 μ l, 16.0 mmol) in freshly distilled THF (20 ml) was added 1 M n -Bu₄NF/THF (6.41 ml, 6.41 mmol). Then the mixture was stirred at room temperature overnight and concentrated in vacuo. The residue was extracted with CHCl₃, successively washed with water and brine, dried over $Na₂SO₄$, and concentrated in vacuo. The crude product was chromatographed on silica gel with hexane/EtOAc (4:1–2:1) to give 11 as a mixture of anomers (1.46 g, 89%, $\alpha/\beta > 10$). R_f 0.50 (1:1 hexane/EtOAc). ¹H NMR: d 7.70–7.69, 7.44–7.11 (m, 30H, Ar), 6.88 (d, 1H, $J=7.8$ Hz, $-NH$), 5.33 (br t, 1H, $J=3.4$ Hz, H-1a), 5.02 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.94 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$, 4.83 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.78 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.71 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.68 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.53 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.50 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.32 (d, 1H, $J=8.1$ Hz, H-1b), 4.30 (d, 1H, $J=12.6$ Hz, $-CH_2Ph$, 4.29 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.19 (d, 1H, $J=11.7$ Hz, $-CH₂Ph$, 4.14–4.08 (m, 1H, H-2a), 4.02–3.93 $(m, 2H, H-4a$ and H-5a), 3.89 (d, 1H, $J=2.4$ Hz, H-4b), 3.82–3.74 (m, 3H, H-2b, H-3a, and H-6a), 3.59 (br d, 1H, J¼9.8 Hz, H-6a), 3.43–3.34 (m, 4H, H-3b, H-5b, H-6b, OH), $3.32-3.28$ (m, 1H, H-6b). ¹³C NMR: δ 161.4 (Cl_3CCONH) , 102.9 (C-1b), 92.5 (– CCl_3), 90.8 (C-1a, α anomer). MALDI TOF MS calcd for $C_{56}H_{58}Cl_3NO_{11}$ m/z: 1025.31. Found: 1048.27 (M+Na⁺). Anal. Calcd for $C_{56}H_{58}Cl_3NO_{11}$: C, 65.46; H, 5.69; N, 1.36. Found: C, 65.73; H, 5.72; N, 1.24.

4.1.4. 2,3,4,6-Tetra-*O*-benzyl-β-D-galactopyranosyl- $(1 \rightarrow$ 4)-3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-b-D-glucopyranosyl fluoride 12. To a stirred solution of 11 (635 mg, 0.62 mmol) in freshly distilled THF (10 ml) was added Et₂NSF₃ (121 ul, 0.93 mmol) at 0 °C. The mixture was stirred for 1 h before the reaction was quenched with MeOH and concentrated in vacuo. The residue was extracted with EtOAc, successively washed with water and brine, dried over $Na₂SO₄$, and concentrated in vacuo. The crude product was chromatographed on silica gel with toluene/EtOAc (9:1) to give 12 as a mixture of anomers (582 mg, 92%, α/β =19/1). R_f 0.40 (9:1 toluene/EtOAc). ¹H NMR: δ 7.36–7.14 (m, 30H, Ar), 6.54 (d, 1H, $J=8.1$ Hz, $-NH$), 5.75 (dd, 1H, $J=2.4$, 53.7 Hz, H-1a), 5.03 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.96 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.84 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$, 4.76 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.72–4.67 (m, 2H, $-CH_2Ph$), 4.66 (d, 1H, J=8.3 Hz, H-1b), 4.63 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.56 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$, 4.54 (d, 1H, J=11.5 Hz, $-CH_2Ph$), 4.38–4.36 (m, 4H, $-CH_2Ph$, 4.24 (d, 1H, $J=11.3$ Hz, $-CH_2Ph$), 4.15 (t, 1H, $J=9.5$ Hz, H-2a), 3.92–3.89 (m, 2H, H-4b and H-5a), 3.86 (br d, 1H, J=11.4 Hz, H-2b), 3.79–3.72 (m, 2H, H-3a and H-6a), 3.56 (dd, 1H, $J=1.2$, 11.0 Hz, H-6a), 3.52–3.43 (m, 1H, H-6b), 3.38–3.32 (m, 3H, H-4a, H-5b, and H-6b). ¹³C NMR: δ 161.5 (Cl₃CCONH), 105.3 (d, J_{CF}=220.1 Hz, C-1a, α -F), 102.4 (C-1b), 91.9 (–CCl₃). MALDI TOF MS calcd for $C_{56}H_{57}Cl_3FNO_{10}$ m/z: 1027.30. Found: 1050.39 $(M+Na^{+})$. Anal. Calcd for $C_{56}H_{57}Cl_{3}FNO_{10}$: C, 65.34; H, 5.58; N, 1.36. Found: C, 65.45; H, 5.58; N, 1.33.

4.1.5. N-(9-Fluorenylmethoxycarbonyl)-O-[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1 \rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1 \rightarrow$ 3)-2-azido-4,6-O-benzylidene-2-deoxy-a-D-galactopyranosyl]-L-serine allyl ester 13. A mixture of Cp_2ZrCl_2 (315 mg, 1.08 mmol), AgClO₄ (447 mg, 2.16 mmol), and

dried molecular sieves 4 Å powder (1 g) in anhydrous CH_2Cl_2 (3 ml) was stirred at room temperature under Ar for 30 min and then cooled at -15 °C. To the mixture was added a mixture of $6 \times (415 \text{ mg}, 0.65 \text{ mmol})$ and 12 $(554 \text{ mg}, 0.54 \text{ mmol})$ in anhydrous CH_2Cl_2 (7 ml). The mixture was stirred at -15 °C for 1 h, before the reaction was quenched with aq NaHCO₃. The mixture was diluted with EtOAc and filtered through Celite. The filtrate was successively washed with satd $NAHCO₃$, water, and brine, dried over $MgSO₄$, and concentrated in vacuo. The residue was chromatographed on silica gel with toluene/EtOAc (88:12) to give 13 (670 mg, 75%). $\lbrack \alpha \rbrack_{D}$ +62.9 (c 1). R_f 0.51 (4:1) toluene/EtOAc). ¹H NMR: δ 7.76 (d, 2H, J=7.6 Hz, Ar), 7.62 (d, 2H, $J=7.3$ Hz, Ar), 7.61-7.06 (m, 40H, Ar and Cl_3CCONH), 5.98–5.85 (m, 2H, – $CH_2CH=CH_2$, FmocNH), 5.351 [s, 1H, PhCH(O)₂], 5.35 (d, 1H, J=17.1 Hz, $-CH=CH_2$), 5.28 (d, 1H, J=10.7 Hz, $-CH=CH_2$), 5.26 (d, 1H, $J=7.1$ Hz, H-1b), 5.04 (d, 1H, $J=10.3$ Hz, $-CH_2Ph$), 4.97 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.95 (d, 1H, $J=3.5$ Hz, H-1a), 4.85 (d, 1H, $J=11.6$ Hz, $-CH₂Ph$), 4.82 (d, 1H, $J=11.6$ Hz, $-CH_2Ph$), 4.74 (d, 1H, $J=12.2$ Hz, $-CH_2Ph$, 4.73-4.68 (m, 1H, $-CH_2Ph$), 4.69 (d, 2H, $J=5.9$ Hz, $-CH_2C=CH_2$), 4.55 (d, 1H, $J=10.3$ Hz, $-CH_2Ph$, 4.54 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 3.92 (d, 1H, $J=2.7$ Hz, H-4c), 3.83 (dd, 1H, $J=3.5$, 10.7 Hz, H-2a), 3.73–3.67 (m, 1H, H-5b), 3.59 (br s, 1H, H-4a), 3.45 (dd, 1H, $J=2.7$, 9.8 Hz, H-3c), 3.39 (dd, 1H, $J=4.9$, 7.6 Hz, H-5c), 3.32 (dd, 1H, J=4.9, 8.5 Hz, H-6c). ¹³C NMR: δ 169.2 (–CO2), 161.7 (Cl3CCONH), 155.7 (OCONH), 131.1 $(-CH=)$, 119.0 $(=CH₂)$, 103.1 $(^1J_{CH}=160.1$ Hz, C-1c), 100.4 [PhCH(O)₂], 99.8 $(^1J_{CH} = 168.4 \text{ Hz}, \text{ C-1b}, \ ^1J_{CH} =$ 171.8 Hz, C-1a), 92.3 ($-CCl_3$). MALDI TOF MS calcd for $C_{90}H_{90}N_5O_{19}Cl_3$ m/z: 1649.53. Found: 1672.56 (M+Na⁺), 1688.34 (M+K⁺). Anal. Calcd for $C_{90}H_{90}N_5O_{19}Cl_3$: C, 65.43; H, 5.49; N, 4.24. Found: C, 65.61; H, 5.64; N, 3.90.

4.1.6. N-(9-Fluorenylmethoxycarbonyl)-O-[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1 \rightarrow$ 3)-2-azido-4,6-O-benzylidene-2-deoxy-a-D-galactopyranosyl]-L-threonine allyl ester 14. Condensation of 7 (368 mg, 0.56 mmol) and 12 (481 mg, 0.47 mmol) was performed in a similar manner as described for 13. Chromatography of the crude product on silica gel with toluene/ EtOAc (88:12) gave 14 (622 mg, 80%). $[\alpha]_D$ +58.8 (c 1). R_f 0.34 (7:1 toluene/EtOAc). ¹H NMR: δ 7.77 (d, 2H, $J=7.6$ Hz, Ar), 7.64 (d, 2H, $J=7.3$ Hz, Ar), 7.52–7.06 (m, 40H, Ar and Cl₃CCONH), 5.92 (ddt, 1H, $J=6.1$, 10.5, 17.1 Hz, $-CH_2CH=CH_2$), 5.72 (d, 1H, $J=9.5$ Hz, FmocNH), 5.36 [s, 1H, PhCH(O)₂], 5.35 (dd, 1H, $J=1.2$, 17.1 Hz, $-CH=CH_2$), 5.33 (d, 1H, J=7.8 Hz, H-1b), 5.25 (dd, 1H, $J=1.2$, 10.5 Hz, $-CH=CH_2$), 5.05 (d, 1H, $J=10.3$ Hz, $-CH_2Ph$, 5.01 (d, 1H, $J=3.7$ Hz, H-1a), 4.97 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.85 (d, 1H, $J=$ 11.2 Hz, $-CH_2Ph$, 4.82 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.74 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.70 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.66 (m, 2H, $-CH_2C=CH_2$), 4.54 (d, 2H, J=11.2 Hz, -CH₂Ph), 4.49-4.39 (m, 4H, Thr- β H, H-6a, and $-OCH₂CHAr₂$), 4.40 (d, 1H, J=7.8 Hz, H-1c), 4.33 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.32 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$, 4.31–4.27 (m, 3H, H-3b, H-6a, and $-CH_2CHAr_2$), 4.22 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$), 4.18 (d, 1H, $J=11.5$ Hz, $-CH_2P$ h), 4.11–4.04 (m, 2H, H-6b and H-3a), 3.92 (d, 1H,

 $J=2.7$ Hz, H-4c), 3.93–3.78 (m, 2H, H-4b and H-5a), 3.85 (dd, 1H, $J=3.7$, 10.7 Hz, H-2a), 3.80 (dd, 1H, $J=7.8$, 9.3 Hz, H-2c), 3.76–3.72 (m, 2H, H-5b and Thr–aH), 3.63 (br d, 1H, $J=12.0$ Hz, H-6b), 3.58 (br s, 1H, H-4a), 3.52– 3.45 (m, 3H, H-2b, H-6c, and H-3c), 3.41 (dd, 1H, $J=5.1$, 7.8 Hz, H-5c), 3.32 (dd, 1H, $J=5.1$, 8.8 Hz, H-6c), 1.30 (d, 3H, J=6.3 Hz, Thr– γ H). ¹³C NMR: δ 169.7 (–CO₂), 161.8 (Cl₃CCONH), 156.6 (OCONH), 131.1 (-CH=), 119.2 $(=CH₂), 103.1$ (¹J_{CH}=160.1 Hz, C-1c), 100.5 [PhCH(O)₂], 99.6 (¹J_{CH}=165.9 Hz, C-1b), 99.3 (¹J_{CH}=173.4 Hz, C-1a), 92.3 ($-CCl_3$). MALDI TOF MS calcd for $C_{91}H_{92}N_5O_{19}Cl_3$ m/z: 1663.55. Found: 1686.85 (M+Na⁺), 1702.75 (M+K⁺). Anal. Calcd for $C_{91}H_{92}N_5O_{19}Cl_3$: C, 65.60; H, 5.57; N, 4.20. Found: C, 65.84; H, 5.74; N, 3.87.

4.1.7. N-(9-Fluorenylmethoxycarbonyl)-O-[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1 \rightarrow$ 3)-2-azido-2-deoxy-a-D-galactopyranosyl]-L-serine allyl ester 15. To an ice-cooled solution of 13 (266 mg, 0.16 mmol) in CH_2Cl_2 (8 ml) was added 80% aq CF_3CO_2H (4 ml) with stirring. The mixture was stirred at 0° C for 30 min, neutralized by addition of satd $NaHCO₃$, and extracted with EtOAc. The extract was successively washed with satd NaHCO₃, water, and brine, dried over $MgSO₄$, and concentrated in vacuo. The crude product was chromatographed on silica gel with toluene/EtOAc (2:1) to give 15 (237 mg, 94%). $[\alpha]_D$ +41.9 (c 1). R_f 0.18 (3:2 toluene/ EtOAc). ¹H NMR: δ 7.74 (d, 2H, J=7.6 Hz, Ar), 7.62 (d, 2H, $J=7.3$ Hz, Ar), $7.57-7.06$ (m, 35H, Ar and Cl₃CCONH), 6.09 (d, 1H, $J=8.3$ Hz, FmocNH), 5.90 (ddt, 1H, $J=5.9$, 10.5, 16.8 Hz, $-CH_2CH=CH_2$), 5.33 (d, 1H, $J=16.8$ Hz, $-CH=CH_2$), 5.25 (1H, d, J=10.5 Hz, $-CH=CH_2$), 5.21 (1H, d, $J=7.8$ Hz, H-1b), 5.02 (1H, d, $J=10.5$ Hz, $-CH_2Ph$, 4.97 (d, 1H, J=11.5 Hz, $-CH_2Ph$), 4.85 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.84 (m, 1H, H-1a), 4.79 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.74 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.71 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$), 4.69–4.63 (m, 2H, $-CH_2CH = CH_2$), 4.55 (d, 1H, J=10.5 Hz, $-CH_2Ph$), 4.54 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.54 (m, 1H, Ser– α H), 4.44–4.28 (m, 6H, H-1c, $-NCO_2CH_2CH$, $-CH_2Ph \times 3$), 4.23 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.10 (br s, 1H, H-4a), 3.34 (dd, 1H, $J=4.9$, 8.5 Hz, H-6c), 3.24 (br s, 1H, $-OH$), 2.41 (br s, 1H, $-OH$). ¹³C NMR: δ 169.3 ($-CO_2$), 161.9 (Cl_3CCONH) , 155.6 (OCONH), 131.1 (-CH=), 118.9 $(=CH_2)$, 103.1 $(^1J_{CH} = 160.9$ Hz, C-1c), 99.3 $(^1J_{CH} =$ 170.9 Hz, C-1a), 98.8 ($^{1}J_{CH}$ =167.6 Hz, C-1b), 92.1 $(-CCl_3)$. MALDI TOF MS calcd for $C_{83}H_{86}N_5O_{19}Cl_3$ m/z: 1561.50. Found: 1584.27 (M+Na⁺), 1600.24 (M+K⁺). Anal. Calcd for $C_{83}H_{86}N_5O_{19}Cl_3$: C, 63.74; H, 5.54; N, 4.48. Found: C, 63.42; H, 5.72; N, 4.09.

4.1.8. N-(9-Fluorenylmethoxycarbonyl)-O-[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1 \rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1 \rightarrow$ 3)-2-azido-2-deoxy-a-D-galactopyranosyl]-L-threonine allyl ester 16. Compound 14 (403 mg, 0.24 mmol) was debenzylidenated in a similar manner as described for 15. The crude product was purified by column chromatography on silica gel with toluene/EtOAc (2:1–3:2) to give 16 (319 mg, 83%). $[\alpha]_D$ +34.5 (c 1). R_f 0.50 (1:1 toluene/ EtOAc). ¹H NMR: δ 7.76 (d, 2H, J=7.3 Hz, Ar), 7.63 (d, $2H, J=7.3$ Hz, Ar), 7.41–7.12 (m, 35H, Ar and Cl₃CCONH), 5.92 (ddt, 1H, $J=5.9$, 10.5, 17.1 Hz, $-CH_2CH=CH_2$), 5.66 (d, 1H, $J=9.3$ Hz, FmocNH), 5.34 (br d, 1H, $J=17.1$ Hz, $-CH=CH_2$), 5.29 (d, 1H, J=7.6 Hz, H-1b), 5.25 (d, 1H, $J=10.5$ Hz, $-CH=CH_2$), 5.02 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$, 4.97 (d, 1H, J=11.5 Hz, $-CH_2Ph$), 4.95 (d, 1H, $J=5.1$ Hz, H-1a), 4.85 (d, 1H, $J=11.0$ Hz, $-CH₂Ph$), 4.79 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$), 4.75 (d, 1H, $J=12.0$ Hz, $-CH_2Ph$, 4.71 (d, 1H, J=12.0 Hz, $-CH_2Ph$), 4.65 (d, 2H, $J=5.9$ Hz, $-CH_2CH=CH_2$), 4.54 (d, 2H, $J=11.2$ Hz, $-CH_2Ph \times 2$), 4.48–4.20 (m, 1H, H-1c), 4.15 (br s, 1H, H-4a), 4.01 (dd, 1H, $J=2.7$, 10.5 Hz, H-3a), 3.94 (d, 1H, $J=2.4$ Hz, H-4c), 3.35 (dd, 1H, $J=4.9$, 8.5 Hz, H-6c), 3.25 (br s, 1H, –OH), 2.30 (br s, 1H, –OH), 1.30 (d, 3H, J=6.3 Hz, Thr– γ H); ¹³C NMR: δ 169.8 (–CO₂), 162.1 $(Cl₃CCONH)$, 156.6 (OCONH), 131.2 (–CH=), 119.2 $(=CH_2)$, 103.2 $(^1J_{CH}=160.1 \text{ Hz}$, C-1c), 98.9 $(^1J_{CH}=$ 175.1 Hz, C-1a), 98.4 $(^1J_{CH} = 161.8 \text{ Hz}$, C-1b), 92.1 $(-CCl_3)$. MALDI TOF MS calcd for $C_{84}H_{88}N_5O_{19}Cl_3$ m/z: 1575.51. Found: 1598.69 (M+Na⁺). Anal. Calcd for C84H88N5O19Cl3: C, 63.94; H, 5.62; N, 4.44. Found: C, 64.11; H, 5.46; N, 4.15.

4.1.9. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1 \rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1 \rightarrow$ 3)-[2,3,4,6-tetra-*O*-benzyl-β-D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-b-D-glucopyranosyl- $(1\rightarrow 6)$]-2-azido-2-deoxy- α -D-galactopyranosyl}-L-threonine allyl ester 17 and N-(9-fluorenylmethoxycarbonyl)-O-{2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-B-p-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-Obenzyl- β -D-galactopyranosyl- $(1 \rightarrow 4)$ -3,6-di-O-benzyl-2deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1\rightarrow 4)$]-2-azido-2-deoxy-a-D-galactopyranosyl}-L-threonine allyl ester 18. Condensation of 12 $(29 \text{ mg}, 28 \text{ µmol})$ and 16 (54 mg, 34 µmol) was performed at -15 °C for 1 h in a similar manner as described for 13. Chromatography of the crude product on silica gel with toluene/EtOAc (2:1) gave 17 (32 mg, 44%) and isomer 18 (22 mg, 30%). Unconsumed 16 (11 mg, 21%) was also recovered.

Compound 17: $[\alpha]_D$ +20.1 (c 1.7). R_f 0.39 (3:1 toluene/ EtOAc). ¹H NMR: δ 7.75 (d, 2H, J=7.6 Hz, Ar), 7.62 (d, 1H, $J=6.8$ Hz, Ar), 7.60 (d, 1H, $J=6.8$ Hz, Ar), 7.45–7.06 (m, 65H, Ar and Cl₃CCONH), 7.01 (d, 1H, $J=7.8$ Hz, Cl_3CCONH , 5.91 (ddd, 1H, $J=17.1$, 10.7, 5.9 Hz, $-CH=CH_2$), 5.61 (d, 1H, J=9.3 Hz, FmocNH), 5.34 (dd, 1H, J=17.1, 1.2 Hz, $-CH=CH_2$), 5.24 (d, 1H, J=10.7 Hz, $-CH=CH_2$), 5.23 (d, 1H, J=6.8 Hz, H-1b), 5.01 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$, 4.97 (d, 1H, $J=11.4$ Hz, $-CH_2Ph$), 4.95 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.88 (d, 1H, $J=7.1$ Hz, H-1d), 4.87 (d, 1H, $J=3.2$ Hz, H-1a), 4.81 (d, 1H, $J=$ 11.2 Hz, $-CH_2Ph$, 4.80 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.78 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 3.02 (br s, 1H, $-OH$), 1.30 (d, 3H, J=6.3 Hz, Thr– γ H). ¹³C NMR: δ 169.9 (–CO₂), 162.1 (Cl₃CCONH), 161.4 (Cl₃CCONH), 156.6 (OCONH), 119.3 (=CH₂), 103.02 (¹J_{CH}=165.9 Hz, C-1c or C-1e), 102.96 $(^1J_{CH} = 165.9$ Hz, C-1c or C-1e), 99.9 $(^1J_{CH} =$ 167.6 Hz, C-1b or C-1d), 99.1 $(^1J_{\text{CH}}=174.3$ Hz, C-1a), 98.6 (${}^{1}J_{\text{CH}}$ =160.9 Hz, C-1b or C-1d), 92.5 (-CCl₃), 92.2 $(-CCl₃)$, 18.8 (Thr- γ C). MALDI TOF MS calcd for $C_{140}H_{144}N_6O_{29}Cl_6$ m/z: 2582.81. Found: 2606.11 (M+Na⁺).

Anal. Calcd for $C_{140}H_{144}N_6O_{29}Cl_6 \cdot H_2O$: C, 64.54; H, 5.65; N, 3.23. Found: C, 64.61; H, 5.67; N, 2.93.

Compound 18: $[\alpha]_D$ +35.5 (c 1.4). R_f 0.35 (3:1 toluene/ EtOAc). ¹H NMR: δ 7.75 (d, 2H, J=7.3 Hz, Ar), 7.64 (d, $2H, J=7.3$ Hz, Ar), 7.48–7.01 (m, 65H, Ar and Cl₃CCONH), 6.96 (d, 1H, J=8.5 Hz, Cl₃CCONH), 5.90 (ddd, 1H, J= 17.1, 10.5, 5.9 Hz, $-CH=CH_2$), 5.64 (d, 1H, $J=9.5$ Hz, FmocNH), 5.33 (d, 1H, J=17.1 Hz, $-CH=CH_2$), 5.32 (d, 1H, $J=8.1$ Hz, H-1b or H-1d), 5.24 (d, 1H, $J=10.5$ Hz, $-CH=CH_2$), 5.14 (d, 1H, J=8.3 Hz, H-1b or H-1d), 5.05 (d, 1H, $J=10.0$ Hz, $-CH_2Ph$), 4.93 (d, 2H, $J=11.2$ Hz, $-CH_2Ph \times 2$), 4.85 (d, 1H, $J=3.7$ Hz, H-1a), 4.81 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$, 4.80 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 1.29 (d, 3H, J=6.3 Hz, Thr– γ H). ¹³C NMR: δ 169.8 $(-CO₂)$, 162.0 (Cl₃CCONH), 161.5 (Cl₃CCONH), 156.7 (OCONH), 119.3 (=CH₂), 103.3 (¹J_{CH}=159.2 Hz, C-1c or C-1e), 102.7 $(^1J_{CH} = 159.2 \text{ Hz}$, C-1c or C-1e), 100.4 $(^1J_{CH} = 162.5$ Hz, C-1b or C-1d), 100.3 $(^1J_{CH} = 162.5$ Hz, C-1b or C-1d), 98.9 (${}^{1}J_{CH}$ =172.5 Hz, C-1a), 92.6 (-CCl₃), 92.2 (–CCl₃), 19.1 (Thr– γ C). MALDI TOF MS calcd for $C_{140}H_{144}N_6O_{29}Cl_6$ m/z: 2582.81. Found: 2606.11 m/z : 2582.81. Found: (M+Na⁺), 2622.05 (M+K⁺).

4.1.10. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-b-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1\rightarrow 6)$]-2-azido-2-deoxy-a-D-galactopyranosyl}-L-serine allyl ester 19. A mixture of Cp_2ZrCl_2 (89 mg, 0.30 mmol), AgClO₄ $(127 \text{ mg}, 0.61 \text{ mmol})$, and dried molecular sieves 4 Å powder (700 mg) in anhydrous CH_2Cl_2 (5 ml) was stirred at room temperature under Ar for 30 min and then cooled at -40 °C. To the stirred mixture was added a mixture of 3 (143 mg, 0.15 mmol) and 15 (218 mg, 0.14 mmol) in anhydrous CH_2Cl_2 (15 ml). Then the temperature was raised to -15 °C for 30 min and stirring was continued for further 1.5 h, before the reaction was quenched with aq NaHCO₃. The mixture was diluted with EtOAc and filtered through Celite. The filtrate was successively washed with satd NaHCO₃, water, and brine, dried over MgSO₄, and concentrated in vacuo. The residue was chromatographed on silica gel with toluene/EtOAc $(4:1 \text{ to } 3:1)$ to give 19 (233 mg) , 68%). Heptasaccharide 23 (20 mg, 4%) and isomer 21 (20 mg, 6%) were obtained in the less polar and the more polar fractions, respectively. Unconsumed 15 (13 mg, 6%) was also recovered. Compound 19: $[\alpha]_D$ +28.5 (c 1). R_f 0.71 (3:2) toluene/EtOAc). ¹H NMR (CDCl₃): δ 7.71 (d, 1H, J=6.6 Hz, Ar), 7.70 (d, 1H, J=6.8 Hz, Ar), 7.61 (d, 1H, J=7.6 Hz, Ar), 7.55 (d, 1H, J=7.3 Hz, Ar), 7.54–7.09 (m, 61H, Ar and $Cl_3CCONH \times 2$, 5.90 (m, 1H, $-CH_2CH=CH_2$), 5.80 (d, 1H, $J=7.3$ Hz, FmocNH), 5.46 [s, 1H, PhCH(O)₂], 5.31 (d, 1H, $J=16.6$ Hz, $-CH=CH_2$), 5.22 (br d, 2H, $J=10.3$ Hz, $-CH=CH_2$, $-CH_2Ph$), 5.13 (d, 1H, $J=7.6$ Hz, H-1b), 5.01 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$), 4.97 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$, 4.95 (d, 1H, $J=7.8$ Hz, H-1d), 4.80 (d, 1H, $J=5.1$ Hz, H-1a), 4.84 (d, 1H, $J=11.5$ Hz, $-CH₂Ph$), 2.98 (br s, 1H, -OH), 2.95 (s, 1H, H-4e). ¹³C NMR: δ 169.0 $(-CO₂)$, 161.8 (Cl₃CCONH), 161.4 (Cl₃CCONH), 155.5 (OCONH), 131.1 ($-CH=$), 118.9 ($=CH₂$), 102.9 ($^1J_{CH}$ 160.1 Hz, C-1c or C-1e), 102.6 $(^1J_{\text{CH}}=160.9$ Hz, C-1c or

C-1e), 101.2 [PhCH(O)₂], 99.3 $(^1J_{CH} = 163.4$ Hz, C-1d), 99.0 (${}^{1}J_{\text{CH}}$ =169.3 Hz, C-1b), 97.5 (${}^{1}J_{\text{CH}}$ =173.4 Hz, C-1a), 92.4 (-CCl₃), 92.1 (-CCl₃). MALDI TOF MS calcd for $C_{132}H_{134}N_6O_{29}Cl_6$ m/z: 2476.73. Found: 2499.82 $(M+Na⁺)$. Anal. Calcd for C₁₃₂H₁₃₄N₆O₂₉Cl₆: C, 63.90; H, 5.44; N, 3.39. Found: C, 63.68; H, 5.47; N, 3.20.

Compound 21: $[\alpha]_D$ +43.7 (c 1.4). R_f 0.36 (3:2 toluene/ EtOAc). ¹H NMR: δ 7.75 (d, 2H, J=7.3 Hz, Ar), 7.63 (d, 1H, J=7.8 Hz, Ar), 7.61 (d, 1H, J=7.8 Hz, Ar), 7.56–7.01 $(m, 61H, Ar and Cl₃CCONH×2), 5.97–5.81$ (m, 2H, $-CH = CH_2$ and FmocNH), 5.43 [s, 1H, PhCH(O)₂], 5.32 (d, 1H, $J=17.1$ Hz, $-CH=CH_2$), 5.25 (br d, 2H, $J=9.3$ Hz, $-CH=CH_2$ and H-1b or H-1d), 5.12 (d, 1H, $J=8.3$ Hz, H-1b or H-1d), 5.05 (d, 1H, $J=10.2$ Hz, $-CH_2Ph$), 5.02 (d, 1H, $J=10.7$ Hz, $-CH_2Ph$), 4.94 (d, 1H, $J=11.7$ Hz, $-CH_2$ Ph), 4.36 (d, 1H, J=8.3 Hz, H-1c or H-1e), 2.99 (br s, 1H, H-4e). ¹³C NMR: δ 169.5 (-CO₂), 162.0 (Cl₃CCONH), 161.6 (Cl₃CCONH), 156.0 (OCONH), 119.2 (=CH₂), 103.4 $(^1J_{\text{CH}}=162.2 \text{ Hz}$, C-1c or C-1e), 102.8 $(^1J_{\text{CH}}=$ 159.7 Hz, C-1c or C-1e), 101.1 [PhCH(O)₂], 100.7 ($^{1}J_{CH}$ = 166.3 Hz, C-1b or C-1d), 100.3 $(^1J_{CH} = 168.8$ Hz, C-1b or C-1d), 99.2 $(^1J_{CH} = 170.5 \text{ Hz}$, C-1a), 92.6 $(-CCl_3)$, 92.3 $(-CCl_3)$. MALDI TOF MS calcd for $C_{132}H_{134}N_6O_{29}Cl_6$ m/z: 2476.73. Found: 2499.74 (M+Na⁺). Anal. Calcd for $C_{132}H_{134}N_6O_{29}Cl_6 \cdot H_2O$: C, 63.44; H, 5.48; N, 3.36. Found: C, 63.68; H, 5.54; N, 3.07.

Compound 23: $[\alpha]_D$ +26.5 (c 0.7). R_f 0.56 (2:1 toluene/ EtOAc). ¹H NMR: δ 7.71 (d, 2H, J=7.3 Hz, Ar), 7.57 (d, 1H, J=7.8 Hz, Ar), 7.54 (d, 1H, J=7.3 Hz, Ar), 7.51–6.97 (m, 88H, Ar and $Cl_3CCONH\times3)$, 5.95–5.76 (m, 2H, $-CH=CH₂$ and FmocNH), 5.43 [s, 2H, PhCH(O)₂×2], 5.30 (d, 1H, $J=17.1$ Hz, $-CH=CH_2$), 5.25–5.09 (m, 5H, $-CH=CH_2$, GlcNTCA H-1×2, $-CH_2Ph \times 2$), 5.04 (d, 1H, $J=10.2$ Hz, $-CH_2Ph$, 4.92 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.87 (d, 1H, $J=2.9$ Hz, H-1a), 2.94 (br s, 1H, H-4e or H-4g), 2.84 (br s, 1H, H-4e or H-4g). ¹³C NMR: δ 169.6 $(-CO₂)$, 162.0 (Cl₃CCONH), 161.8 (Cl₃CCONH), 161.4 $(Cl₃CCONH)$, 156.0 (OCONH), 119.2 (=CH₂), 103.1 $(^1J_{\text{CH}}=159.7 \text{ Hz})$, 102.8 $(^1J_{\text{CH}}=163.9 \text{ Hz})$, and 102.5 $(^1J_{CH} = 159.7 \text{ Hz})$ (C-1c, C-1e, and C-1g), 101.3 [PhCH(O)₂], 101.26 [PhCH(O)₂], 100.9 (¹J_{CH}=158.9 Hz), 99.8 (¹J_{CH}= 162.2 Hz), and 99.4 $(^1J_{\text{CH}}=164.7 \text{ Hz})$ (C-1b, C-d, and C-1f), 98.0 $(^1J_{CH} = 170.5 \text{ Hz}$, C-1a), 92.7 $(-CCl_3)$, 92.6 $(-CCl₃), 92.3 (-CCl₃). MALDI TOF MS calcd for$ $C_{182}H_{184}N_7O_{39}Cl_9$ m/z: 3391.97. Found: 3431.09 (M+K⁺). Anal. Calcd for $C_{181}H_{182}N_7O_{39}Cl_9 \tcdot 2H_2O$: C, 63.30; H, 5.46; N, 2.85. Found: C, 63.21; H, 5.40; N, 2.62.

4.1.11. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido-β-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -3,6-di-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-glucopyranosyl- $(1\rightarrow 6)$]-2-azido-2-deoxya-D-galactopyranosyl}-L-threonine allyl ester 20. In a similar manner as described for 19, condensation of 3 (204 mg, 0.22 mmol) and 16 (312 mg, 0.20 mmol) was performed with Cp_2ZrCl_2 (127 mg, 0.43 mmol), AgClO₄ (181 mg, 0.87 mmol), and dried molecular sieves 4 Å powder (1 g) in anhydrous CH_2Cl_2 (30 ml) at -40 to -15 °C for 2 h. Chromatography of the crude product on silica gel with

toluene/EtOAc $(88:12 \text{ to } 80:20)$ afforded 20 (351 mg) 71%), 22 (43 mg, 9%), 24 (55 mg, 8%), and unreacted 16 (13 mg, 4%).

Compound 20: $[\alpha]_D$ +21.7 (c 1). R_f 0.54 (2:1 toluene/ EtOAc). ¹H NMR: δ 7.74 (d, 2H, J=7.3 Hz, Ar), 7.60 (t, 2H, $J=7.3$ Hz, Ar), 7.60–7.05 (m, 60H, Ar and Cl₃CCONH), 6.95 (d, 1H, $J=7.3$ Hz, Cl₃CCONH), 5.90 (m, 1H, $-CH_2CH = CH_2$), 5.59 (d, 1H, J=9.5 Hz, FmocNH), 5.45 [s, 1H, PhCH(O)₂], 5.33 (d, 1H, J=17.1 Hz, –CH=CH₂), 5.24 (d, 1H, $J=10.3$ Hz, $-CH=CH_2$), 5.23 (d, 1H, $J=$ 7.6 Hz, H-1b), 5.19 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$), 5.00 (d, 1H, $J=10.5$ Hz, $-CH_2Ph$), 4.96 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$, 4.88 (d, 1H, J=8.8 Hz, H-1d), 4.87 (d, 1H, $J=4.1$ Hz, H-1a), 4.84 (d, 1H, $J=11.5$ Hz, $-CH₂Ph$), 4.49– 4.28 (m, 10H, H-1c, H-1e, Thr– α H, $-CH_2CHAr_2$, Thr– β H, and $-CH_2Ph \times 4$), 2.99 (br s, 1H, $-OH$), 2.90 (s, 1H, H-4e), 1.29 (d, 3H, J=6.3 Hz, Thr- γ H). ¹³C NMR: δ 169.6 $(-CO₂)$, 161.9 $(Cl₃CCONH)$, 161.3 $(Cl₃CCONH)$, 156.5 (OCONH), 131.1 ($-CH=$), 119.1 ($=CH₂$), 102.9 $(^1J_{CH} = 160.1 \text{ Hz}$, C-1c or C-1e), 102.5 $(^1J_{CH} = 160.1 \text{ Hz}$, C-1c or C-1e), 101.2 [PhCH(O)₂], 99.9 ($^{1}J_{\text{CH}}$ =165.9 Hz, C-1d), 99.0 (${}^{1}J_{CH}$ =172.5 Hz, C-1a), 98.6 (${}^{1}J_{CH}$ =169.2 Hz, C-1b), 92.4 $(-CCl_3)$, 92.1 $(-CCl_3)$. MALDI TOF MS calcd for $C_{133}H_{136}N_6O_{29}Cl_6$ m/z: 2490.75. Found: 2514.13 $(M+Na⁺)$. Anal. Calcd for C₁₃₃H₁₃₆N₆O₂₉Cl₆: C, 64.02; H, 5.49; N, 3.37. Found: C, 64.02; H, 5.32; N, 3.22.

Compound 22: $[\alpha]_D$ +38.0 (c 1.8, CHCl₃). R_f 0.21 (2:1 toluene/EtOAc). ¹H NMR: δ 7.75 (d, 2H, $J=7.3$ Hz, Ar), 7.63 (d, 2H, $J=7.3$ Hz, Ar), $7.57-7.00$ (m, 61H, Ar and $Cl_3CCONH \times 2)$, 5.90 (ddd, 1H, J=16.6, 10.7, 5.9 Hz, $-CH=CH_2$), 5.67 (d, 1H, J=9.8 Hz, FmocNH), 5.43 [s, 1H, PhCH(O)₂], 5.33 (d, 1H, J=16.6 Hz, $-CH=CH_2$), 5.30 (d, 1H, $J=7.8$ Hz, H-1b or H-1d), 5.24 (d, 1H, $J=10.2$ Hz, $-CH_2Ph$, 5.13 (d, 1H, $J=8.3$ Hz, H-1b or H-1d), 5.06 (d, 1H, $J=10.2$ Hz, $-CH_2Ph$), 5.00 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.93 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$), 4.85 (d, 1H, $J=3.4$ Hz, H-1a), 4.81 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.80 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$), 4.78 (d, 1H, $J=11.2$ Hz, $-CH_2Ph$, 2.98 (br s, 1H, H-4e), 1.29 (d, 3H, J=6.3 Hz, Thr– γ H). ¹³C NMR: δ 170.0 (–CO₂), 162.1 (Cl_3CCONH) , 161.6 (Cl_3CCONH) , 156.8 $(OCONH)$, 119.4 (=CH₂), 103.4 (¹J_{CH}=158.9 Hz, C-1c or C-1e), 102.8 ($^{1}J_{CH}$ =158.9 Hz, C-1c or C-1e), 101.1 [PhCH(O)₂], 100.5 $(^1J_{\text{CH}}=169.6 \text{ Hz}$, C-1b or C-1d), 100.4 $(^1J_{\text{CH}}=$ 169.6 Hz, C-1b or C-1d), 99.1 $(^1J_{CH} = 173.0$ Hz, C-1a), 92.6 ($-CCl_3$), 92.2 ($-CCl_3$), 18.9 (Thr $-\gamma C$). MALDI TOF MS calcd for $C_{133}H_{136}N_6O_{29}Cl_6$ m/z: 2490.75. Found: 2513.67 (M+Na⁺). Anal. Calcd for C₁₃₃H₁₃₆N₆O₂₉Cl₆: C, 64.02; H, 5.49; N, 3.37. Found: C, 63.89; H, 5.58; N, 3.18.

Compound 24: $[\alpha]_D$ +25.9 (c 1.0). R_f 0.64 (2:1 toluene/ EtOAc). ¹H NMR: δ 7.73 (d, 2H, J=7.3 Hz, Ar), 7.61 (d, 1H, J=7.3 Hz, Ar), 7.57 (d, 1H, J=9.3 Hz, Ar), 7.55–6.99 (m, 87H, Ar and Cl₃CCONH \times 2), 6.97 (d, 1H, J=8.3 Hz, Cl₃CCONH), 5.95–5.81 (m, 1H, $-CH=CH_2$), 5.59 (d, 1H, $J=9.8$ Hz, FmocNH), 5.43 [s, 1H, PhCH(O)₂], 5.42 [s, 1H, PhCH(O)₂], 5.31 (d, 1H, J=16.6 Hz, –CH=CH₂), 5.27– 5.08 (m, 5H, $-CH=CH_2$, GlcNTCA H-1×2, $-CH_2Ph \times 2$), 5.04 (d, 1H, $J=10.2$ Hz, $-CH_2Ph$), 4.92 (d, 1H, $J=11.2$ Hz, $-CH_2$ Ph), 4.87 (d, 1H, J=2.9 Hz, H-1a), 2.96 (br s, 1H, H-4c, H-4e, or H-4g), 2.80 (br s, 1H, H-4c, H-4e, or H-4g), 1.27 (d, 3H, J=7.3 Hz, Thr– γ H). ¹³C NMR: δ 169.9 $(-CO₂)$, 162.1 (Cl₃CCONH), 161.6 (Cl₃CCONH), 161.4 (Cl_3CCONH) , 156.7 (OCONH), 119.6 (=CH₂), 103.0 $(^1J_{\text{CH}}=158.1 \text{ Hz})$, 102.8 $(^1J_{\text{CH}}=162.2 \text{ Hz})$, and 102.5 $(^1J_{CH} = 160.6 \text{ Hz})$ (C-1c, C-1e, and C-1g), 101.3 $[PhCH(O)_2]$, 101.2 $[PhCH(O)_2]$, 100.8 $(^1J_{CH} = 163.9 \text{ Hz})$, 100.2 (${}^{1}J_{CH}$ =164.7 Hz), and 99.4 (${}^{1}J_{CH}$ =165.5 Hz) (C-1b, C-1d, and C-1f), 99.0 $(^1J_{CH} = 172.1$ Hz, C-1a), 92.6 $(-CCl_3)$, 92.6 $(-CCl_3)$, 92.3 $(-CCl_3)$, 18.6 (Thr– γ C). MALDI TOF MS calcd for $C_{182}H_{184}N_7O_{39}Cl_9$ m/z: 3405.98. Found: 3444.23 (M+K⁺). Anal. Calcd for $C_{182}H_{184}N_7O_{39}Cl_9$: C, 64.06; H, 5.44; N, 2.87. Found: C, 63.97; H, 5.40; N, 2.70.

4.1.12. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy-b-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy- β -Dglucopyranosyl- $(1\rightarrow 6)$]-2-acetamido-2-deoxy- α -D-galactopyranosyl}-L-serine allyl ester 25. A mixture of 19 (254 mg, 102.4 μ mol), powdered Zn (1 g, 15.3 mmol), and AcOH (1 ml, 17.5 mmol) in EtOAc (10 ml) was placed in a round-bottom flask equipped with a reflux condenser. The atmosphere was replaced with a balloon of Ar. The reaction mixture was stirred under microwave irradiation at 150 W for 30 min. The microwave machine was controlled so as to continuously irradiate the flask during this period. The mixture was diluted with $CHCl₃$ and filtered through Celite. The filtrate was successively washed with satd NaHCO₃, water, and brine, dried over $Na₂SO₄$, and concentrated in vacuo. The residue was dissolved in a mixture of $CH₂Cl₂$ (8 ml) and MeOH (2 ml), and stirred with Ac₂O (1 ml) at room temperature for 1 h. The mixture was concentrated in vacuo to the residue, which was chromatographed on silica gel with toluene/EtOAc (1:4) and then with CHCl₃/MeOH (95:5) to afford 25 (200 mg, 85%). $[\alpha]_D$ +26.7 (c 1.1). R_f 0.59 (9:1 CHCl₃/MeOH). ¹H NMR (DMSO- d_6): δ 7.85 (d, 3H, J=7.8 Hz, Ar), 7.75 (d, 2H, J=8.3 Hz, Ar), 7.70–7.67 (m, 2H, Ar), 7.56–7.02 (m, 60H, Ar and NH \times 4), 5.87 (m, 1H, -CH₂CH=CH₂), 5.64 [s, 1H, PhCH(O)₂], 5.29 (d, 1H, J=17.3 Hz, $-CH=CH_2$), 5.18 (d, 1H, $J=10.5$ Hz, $-CH=CH_2$), 5.09 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$, 4.88 (d, 1H, $J=10.7$ Hz, $-CH_2Ph$), 4.84 (d, 1H, J=11.7 Hz, $-CH_2Ph$), 1.83 (s, 3H, Ac), 1.77 (s, 6H, Ac \times 2). ¹³C NMR (CDCl₃): δ 170.5, 170.1, 170.0, 169.7 (CH₃CONH \times 3, $-CO₂$ All), 155.8 (OCONH), 131.1 $(-CH=)$, 118.8 (=CH₂), 102.8 (¹J_{CH}=160.9 Hz, C-1c or C-1e), 102.5 ($^{1}J_{CH}$ =165.1 Hz, C-1c or C-1e), 101.0 [PhCH(O)₂], 100.3 $($ ¹J_{CH}=165.9 Hz, C-1b and C-1d), 97.8 $(^1J_{CH} = 172.6$ Hz, C-1a), 23.41, 23.36, 23.2 (CH₃CO×3). MALDI TOF MS calcd for $C_{134}H_{144}N_4O_{30}$ m/z: 2288.99. Found: 2312.28 (M+Na⁺), 2328.31 (M+K⁺). Anal. Calcd for $C_{134}H_{144}N_4O_{30} \cdot H_2O$: C, 69.71; H, 6.37; N, 2.43. Found: C, 69.78; H, 6.41; N, 2.37.

4.1.13. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy-b-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy- β -Dglucopyranosyl- $(1\rightarrow 6)$]-2-acetamido-2-deoxy- α -D-galactopyranosyl}-L-threonine allyl ester 26. Reduction of 20 (46 mg, $18.4 \text{ }\mu\text{mol}$) was performed with Zn (180 mg, 2.75 mmol) and AcOH (0.2 ml, 3.49 mmol) in EtOAc (2 ml) under microwave irradiation for 30 min, according to the procedure described above. The crude product was acetylated and purified by column chromatography. Toluene/ EtOAc $(1:4)$ followed by CHCl₃/MeOH $(95:5)$ eluted 26 (33 mg, 80%). $[\alpha]_D$ +29.5 (c 1). R_f 0.61 (9:1 CHCl₃/ MeOH). ¹H NMR (DMSO- d_6): δ 7.91–7.05 (m, 67H, Ar and NH \times 4), 5.87 (m, 1H, -CH₂CH=CH₂), 5.64 [s, 1H, PhCH(O)₂], 5.33 (d, 1H, J=16.8 Hz, –CH=CH₂), 5.22 (d, 1H, $J=10.5$ Hz, $-CH=CH_2$), 4.89 (d, 1H, $J=12.9$ Hz, $-CH_2Ph$, 4.85 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$), 4.84 (d, 1H, $J=11.0$ Hz, $-CH_2Ph$, $4.81-4.73$ (m, 4H, H-1b and $-CH_2Ph \times 3$), 4.70 (d, 1H, $J=12.2$ Hz, $-CH_2Ph$), 4.69 (d, 1H, $J=11.2$ Hz, $-CH_2Ph \times 2$), 1.89 (s, 3H, Ac), 1.81 (s, 3H, Ac), 1.80 (s, 3H, Ac), 1.13 (d, 3H, $J=4.9$ Hz, Thr- γ H). ¹³C NMR (CDCl₃): δ 170.6, 170.34, 170.26, 169.8 $(CH_3CONH \times 3, -CO_2$ All), 156.3 (OCONH), 130.9 $(-CH=)$, 119.3 (=CH₂), 103.0 (¹J_{CH}=160.9 Hz, C-1c or C-1e), 102.7 ($^{1}J_{CH}$ =165.1 Hz, C-1c or C-1e), 101.1 [PhCH(O)₂], 100.5 (¹J_{CH}=163.4 Hz, C-1b or C-1d), 100.3 $(^1J_{\text{CH}}=167.6 \text{ Hz}$, C-1b or C-1d), 99.0 $(^1J_{\text{CH}}=170.1 \text{ Hz}$, C-1a), 23.6, 23.4 ($CH_3CO \times 3$), 18.5 (Thr C-4). MALDI TOF MS calcd for $C_{135}H_{146}N_4O_{30}$ m/z: 2303.00. Found: 2325.96 (M+Na⁺), 2341.92 (M+K⁺). Anal. Calcd for $C_{135}H_{146}N_4O_{30} \cdot H_2O$: C, 69.81; H, 6.42; N, 2.43. Found: C, 69.84; H, 6.16; N, 2.30.

4.1.14. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy-b-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy- β p -glucopyranosyl- $(1\rightarrow 6)$]-2-acetamido-2-deoxy- α - p galactopyranosyl}-L-serine 1. A mixture of 25 (49 mg, 0.02 mmol), 5,5-dimethyl-1,3-cyclohexanedione (60 mg, 0.43 mmol), and $Pd(PPh₃)₄$ (3 mg, 2.6 µmol) in freshly distilled THF (5 ml) was stirred under Ar at room temperature for 30 min, before concentrated in vacuo. The residue was chromatographed on silica gel with $CHCl₃/MeOH$ (95:5) to elute less polar 5,5-dimethyl-1,3-cyclohexanedione and the byproducts. Then addition of AcOH (2%) to the eluant afforded 1 (44 mg, 93%). $[\alpha]_D$ +31.1 (c 1). R_f 0.46 (9:1) CHCl₃/MeOH, 2% AcOH). ¹H NMR (DMSO- d_6): δ 7.85 (d, 2H, $J=7.6$ Hz, Ar), $7.80-7.03$ (m, 65H, Ar and $NH \times 4$), 5.63 [s, 1H, PhCH(O)₂], 5.09 (d, 1H, J=10.7 Hz, $-CH_2Ph$, 4.87 (d, 1H, J=11.0 Hz, $-CH_2Ph$), 4.84 (d, 1H, $J=11.7$ Hz, $-CH₂Ph$, 1.85 (s, 3H, Ac), 1.79 (s, 3H, Ac), 1.77 (s, 3H, Ac). ¹³C NMR (DMSO- d_6): δ 171.5 (-CO₂H), 169.0, 168.8, 168.7 (CH₃CONH×3), 155.8 (OCONH), 102.0 (C-1c and C-1e), 101.8, 101.4 (C-1b and C-1d), 99.7 $[PhCH(O)_2]$, 97.7 (C-1a), 23.0 (CH₃CO \times 3). MALDI TOF MS calcd for $C_{131}H_{140}N_4O_{30}$ m/z: 2248.96. Found: 2271.87 (M+Na⁺), 2287.85 (M+K⁺). Anal. Calcd for $C_{131}H_{140}N_4O_{30}H_2O$: C, 69.36; H, 6.31; N, 2.47. Found: C, 69.34; H, 6.33; N, 2.47.

4.1.15. N-(9-Fluorenylmethoxycarbonyl)-O-{2,3-di-Obenzyl-4,6-O-benzylidene- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy-b-D-glucopyranosyl- $(1\rightarrow 3)$ -[2,3,4,6-tetra-O-benzyl- β -D-galactopyranosyl- $(1\rightarrow 4)$ -2-acetamido-3,6-di-O-benzyl-2-deoxy- β -Dglucopyranosyl- $(1\rightarrow 6)$]-2-acetamido-2-deoxy- α -D-galactopyranosyl}-L-threonine 2. Compound 26 (102 mg,

44.3 mmol) was deallylated according to the procedure described for the synthesis of 1. Chromatography of the crude product gave 2 (92 mg, 92%). $[\alpha]_D$ +32.2 (c 1). R_f 0.48 (9:1) CHCl₃/MeOH, 2% AcOH). ¹H NMR (DMSO- d_6): δ 7.93– 7.77, 7.77–7.55, 7.48–7.03 (m, 67H, Ar and NH \times 4), 5.64 [s, 1H, PhC $H(O)_2$], 5.08 (d, 1H, J=10.7 Hz, -CH₂Ph), 4.88 (d, 1H, $J=10.7$ Hz, $-CH_2Ph$), 4.84 (d, 1H, $J=11.7$ Hz, $-CH_2Ph$, 4.82–4.73 (m, 5H, H-1b, and $-CH_2Ph \times 4$), 4.70 (d, 1H, $J=11.5$ Hz, $-CH_2Ph$), 4.69 (d, 3H, $J=12.7$ Hz, $-CH_2Ph\times3$), 4.63 (d, 1H, J=12.5 Hz, $-CH_2Ph$), 4.61 (d, 2H, $J=12.0$ Hz, $-CH_2Ph \times 2$), 1.91 (s, 3H, Ac), 1.794 and 1.790 (2s, $3H \times 2$, Ac $\times 2$), 1.11 (d, 3H, J=6.1 Hz, Thr– γ H). ¹³C NMR (DMSO- d_6): δ 171.4 (–CO₂H), 169.0, 168.8, and 168.4 (CH₃CONH \times 3), 156.3 (OCONH), 102.0 (¹J_{CH}= 161.8 Hz, C-1c and C-1e), 101.3 $(^1J_{CH} = 157.6$ Hz, C-1b and C-1d), 99.7 [PhCH(O)₂], 98.5 ($^1J_{CH}$ =168.4 Hz, C-1a), 23.03, and 22.97 ($CH_3CO \times 3$), 18.8 (Thr C-4). MALDI TOF MS calcd for $C_{132}H_{142}N_4O_{30}$ m/z: 2262.97. Found: 2285.75 (M+Na⁺), 2301.72 (M+K⁺). Anal. Calcd for $C_{131}H_{140}N_4O_{30}\cdot H_2O$: C, 69.46; H, 6.36; N, 2.45. Found: C, 69.22; H, 6.22; N, 2.35.

4.2. Synthesis of glycopeptide 27

Commercial Fmoc-CLEAR-amide resin (435 mg, 0.2 mmol) was stirred with 20% piperidine/NMP (5 ml) in a polypropylene tube by a vortex mixer for 5 min. After filtration the resin was again stirred with 20% piperidine/NMP (5 ml) for 15 min to complete N-deprotection. Then the resin was repeatedly washed with NMP to remove piperidine and reacted for 1 h with $Fmoc-Gly-SCH₂CH₂COOH$ prepared from the carboxylic acid (154 mg, 0.4 mmol) by activation with 1 M DCC in NMP $(0.4 \text{ ml}, 0.4 \text{ mmol})$ and 1 M HOBt in NMP (0.4 ml, 0.4 mmol). The Fmoc group was removed by stirring with a mixture of 1-methylpyrrolidine (25%), hexamethyleneimine (2%), and 1-hydroxybenzotriazole (4.8%) in NMP/ DMSO (1:1, 3 ml) for 2 min and again with the same amount of the mixture for 15 min. After washing the resin with NMP, Tsoc–Lys (Z) –OPfp $(259 \text{ mg}, 0.4 \text{ mmol})$ in THF (2.5 ml) was reacted with it for 30 min. The coupling reaction was repeated with the same amount of Tsoc–Lys(Z)–OPfp. After washing with THF and dichloromethane, Fmoc–Leu–F (142 mg, 0.4 mmol) dissolved in CH_2Cl_2 (2.5 ml) was added to the resin. Then 1 M Bu₄NF in THF (20 μ 1, 0.02 mmol) was added and the resin was stirred for 1 h before washing with THF and NMP. The resulting tripeptide-resin was subjected to the automated synthesis by ABI 433A peptide synthesizer using FastMoc protocol. The Fmoc deprotection protocol was modified so that the premixed reagent was introduced to the reaction vessel without dilution. The deprotection time was 5 and 20 min. After the synthesis of tetradecapeptide $(Asp⁴⁵-Gly⁵⁸)$ was completed, an eighth part of the resin placed in a propylene tube was stirred with the Fmoc deblocking reagent (1 ml) for 5 min and then for 20 min as already mentioned. The resin was washed with NMP and reacted with $2(113 \text{ mg}, 50 \text{ µmol})$, 1 M DCC in NMP (100 μ l, 100 μ mol), and 1 M HOBt in NMP (100 μ l, 100 μ mol) in NMP (75 μ l) at 50 °C for 3 h by using a vortex mixer. The remaining sequence was also manually introduced using Fmoc amino acid (100 µmol), DCC (150 μ mol), and HOBt (150 μ mol) in NMP to give glycopeptide (Gly³⁴-Gly⁵⁸)-resin (108 mg). The experiment for deprotection was performed with the resin (27 mg), which was treated with the Fmoc deblocking reagent as mentioned

above. The resulting N-deprotected glycopeptide-resin was stirred with reagent K (TFA/phenol/water/thioanisole/ ethanedithiol, $33:2:2:2:1$, 500 µl by the vortex mixer at room temperature for 2 h. The volatile materials in the mixture were evaporated in a stream of $N₂$. Ether was added to the residue to precipitate the product, which was separated by centrifugation. The precipitate was washed several times by suspending in ether and then centrifuging to give a crude product, which was dissolved in a mixture of TFA/DMS/ *m*-cresol (5:3:1, 180 μ I) and cooled at -15 °C. TfOH (20 ml) was added to the mixture and the plastic vessel was shaken in the cooling bath. The reaction mixture was left at -15 °C for 2 h, before the reaction was terminated by the addition of ether. The mixture was centrifuged to precipitate the debenzylated product, which was washed three times with ether as mentioned above. The crude product was purified by HPLC to give 27 (0.14 µmol, 2.2% overall yield based on the amino group on the initial resin). MALDI TOF MS calcd for C₁₅₈H₂₆₂N₃₈O₆₃S₂ m/z: 3763.79. Found: 3764.95 $(M+H⁺)$. Amino acid analysis: Asp_{1.00}Thr_{3.26}Ser_{2.42}Glu_{0.94}- $Gly_3Ala_{1.14}Val_{1.36}Ile_{0.94}Leu_{3.74}Lys_{2.01}His_{0.97}Arg_{1.25}$.

Acknowledgements

This work was supported by grant-in-aid for creative scientific research (17GS0420) from Japan Society for the Promotion of Science and in part by NEDO (New Energy and Industrial Technology Development Organization, Japan). We thank Dr. T. Chihara and his staff at Riken for the combustion analyses. We also thank Tokai University for support by grant-in-aid for high-technology research.

References and notes

- 1. Varki, A. Essentials of Glycobiology; Varki, A., Cummings, R., Esko, J., Freeze, H., Hart, G., Marth, J., Eds.; Cold Spring Harbor Laboratory: New York, NY, 1999; pp 537–549.
- 2. Nakahara, Y. Trends Glycosci. Glycotechnol. 2003, 15, 257– 273.
- 3. Nakahara, Y.; Ozawa, C.; Tanaka, E.; Ohtsuka, K.; Takano, Y.; Hojo, H.; Nakahara, Y. Tetrahedron 2007, 63, 2161–2169.
- 4. (a) Van Halbeek, H.; Dorland; Vliegenthart, J. F. G.; Hull, W. E.; Lamblin, G.; Lhermitte, M.; Boersma, A.; Roussel, P. Eur. J. Biochem. 1982, 127, 7–20; (b) Lamblin, G.; Boersma,

A.; Klein, A.; Roussel, P.; Van Halbeek, H.; Vliegenthart, J. F. G. J. Biol. Chem. 1984, 259, 9051–9059.

- 5. Capon, C.; Laboisse, C. L.; Wieruszeski, J.-M.; Maoret, J.-J.; Augeron, C.; Fournet, B. J. Biol. Chem. 1992, 267, 19248– 19257.
- 6. Hounsell, E. F.; Lawson, A. M.; Stoll, M. S.; Kane, D. P.; Cashmore, G. C.; Carruthers, R. A.; Feeney, J.; Feizi, T. Eur. J. Biochem. 1989, 186, 597–610.
- 7. Hounsell, E. F.; Wood, E.; Feizi, T. Carbohydr. Res. 1981, 90, 283–307.
- 8. Mutsaers, J. H. G.; Van Halbeek, H.; Vliegenthart, J. F. G.; Wu, A. M.; Kabat. Eur. J. Biochem. 1986, 157, 139–146.
- 9. For the reported synthetic core $4 O$ -glycans, see: (a) Mathieux, N.; Paulsen, H.; Meldal, M.; Bock, K. J. Chem. Soc., Perkin Trans. 1 1997, 2359–2368; (b) Misra, A. K.; Ujita, M.; Fukuda, M.; Hindsgaul, O. Carbohydr. Lett. 2001, 4, 71–76.
- 10. (a) Singh, L.; Nakahara, Y.; Ito, Y.; Nakahara, Y. Tetrahedron Lett. 1999, 40, 3769–3772; (b) Singh, L.; Nakahara, Y.; Ito, Y.; Nakahara, Y. Carbohydr. Res. 2000, 325, 132–142.
- 11. (a) Takano, Y.; Habiro, M.; Someya, M.; Hojo, H.; Nakahara, Y. Tetrahedron Lett. 2002, 43, 8395-8399; (b) Takano, Y.; Kojima, N.; Nakahara, Y.; Hojo, H.; Nakahara, Y. Tetrahedron 2003, 59, 8415–8427.
- 12. Suzuki, K.; Maeta, H.; Matsumoto, T. Tetrahedron Lett. 1989, 30, 4853–4856.
- 13. Macindoe, W.; Iijima, H.; Nakahara, Y.; Ogawa, T. Carbohydr. Res. 1995, 269, 227–257.
- 14. Watabe, J.; Singh, L.; Nakahara, Y.; Ito, Y.; Hojo, H.; Nakahara, Y. Biosci. Biotechnol. Biochem. 2002, 66, 1904–1914.
- 15. Lidström, P.; Tierney, J.; Wathey, B.; Westman, J. Tetrahedron 2001, 57, 9225–9283.
- 16. (a) Hojo, H.; Watabe, J.; Nakahara, Y.; Nakahara, Y.; Ito, Y.; Nabeshima, K.; Toole, B. P. Tetrahedron Lett. 2001, 42, 3001–3004; (b) Hojo, H.; Haginoya, E.; Matsumoto, Y.; Nakahara, Y.; Nabeshima, K.; Toole, B. P.; Watanabe, Y. Tetrahedron Lett. 2003, 44, 2961–2964; (c) Haginoya, E.; Hojo, H.; Nakahara, Y.; Nakahara, Y.; Nabeshima, K.; Toole, B. P.; Watanabe, Y. Biosci. Biotechnol. Biochem. 2006, 70, 1338–1349.
- 17. Li, X.; Kawakami, T.; Aimoto, S. Tetrahedron Lett. 1998, 39, 8669–8672.
- 18. King, D. S.; Fields, C. G.; Fields, G. B. Int. J. Pept. Protein Res. 1990, 36, 255–266.
- 19. Tam, J. P.; Heath, W. F.; Merrifield, R. B. J. Am. Chem. Soc. 1986, 108, 5242–5251.